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Stereoselective preparation of 7-exo-amino-7-endo-substituted bicyclo[4.1.0]heptanes

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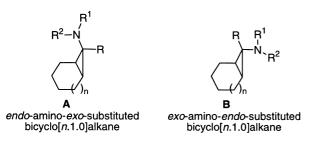
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Abstract—Bicyclo[4.1.0]heptan-7-one N,O-hemiacetals undergo substitution reaction with Wittig reagents (Ph₃P=CHR³) to give stereoselectively the corresponding 7-exo-amino-7-endo-substituted bicyclo[4.1.0]heptanes in good yields. © 2002 Elsevier Science Ltd. All rights reserved.

1. Introduction

Cyclopropanone *N,O*-hemiacetals are versatile precursors for various organic syntheses.¹ It is known that the hydroxy groups of *N,O*-hemiacetals are readily displaced by various nucleophiles such as alcohols, amines, thiols, hydrogen cyanide² and hydrazoic acid.³ Indole, *N*-methylpyrrole, silyl enol ethers or active methylene compounds in the presence of titanium tetrachloride,⁴ Grignard reagents⁴ and C–H acidic compounds^{5–8} also work as nucleophiles to give the corresponding displacement products of the hydroxy groups from *N,O*-hemiacetals. In the case of ring-fused cyclopropanone *N,O*-acetals, *endo*-amino-*exo*-substituted bicyclo[*n*.1.0]alkanes **A** are generally formed as the predominant products, because the displacement involves the cyclopropaniminium cation as an intermediate, which is attacked by nucleophiles from a less-hindered *exo*-position



Scheme 1.

Keywords: bicyclo[4.1.0]heptane; bicyclo[4.1.0]heptan-7-one *N,O*-hemiacetal; Wittig reagent; substitution; *7-endo*-amino-7-*exo*-substituted bicyclo[4.1.0]heptane; *7-exo*-amino-7-*endo*-substituted bicyclo[4.1.0]heptane.

(Scheme 1). 9,10 On the other hand, there have been no reports on stereoselective preparation of the corresponding *exo*-amino-*endo*-substituted bicyclo[n.1.0]alkanes **B** (Scheme 1).

In this paper, we report a convenient method for preparing 7-exo-amino-7-endo-substituted bicyclo[4.1.0]heptanes by the reaction of bicyclo[4.1.0]heptan-7-one *N*,*O*-hemiacetals, which are readily prepared by our procedure, 11,12 with monosubstituted Wittig reagents.

2. Results and discussion

The reaction of bicyclo[4.1.0]heptan-7-one *N,O*-hemiacetals **1a–e** with one equivalent of triphenylphosphorane in the presence of a catalytic amount of benzoic acid¹³ in toluene under reflux conditions predominantly gave the corresponding 7-exo-amino-7-endo-substituted bicyclo-[4.1.0]heptanes **2–5** in 52–87% yields (Scheme 2). In most cases, a small amount of ring-fused alkylidenecyclo-propanes **6** was detected as a by-product.¹⁴ The results are summarized in Table 1.

The reaction of bicyclo[4.1.0]heptan-7-one N,O-hemiacetal **1a** with a Wittig reagent (R³=COOMe) was monitored by GLC analysis, and the results are shown in Fig. 1. A peak of ring-fused alkylidenecyclopropane **6** (R³=COOMe) appeared as a major product in the initial stage of the reaction, and its yield reached about 75% within a few minutes after the start of the reaction. However, we failed to obtain **6** in a reasonable yield, and **2a** was obtained as a final and major product. These results indicate that, in the present reaction, **6** was formed initially and then the liberated amine immediately attacked the β -carbon of α,β -unsaturated ester **6** to give **2a** predominantly. It has already been

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Scheme 2.

Table 1. Reactions of bicyclo[4.1.0]heptan-7-one *N,O*-hemiacetals **1** with Wittig reagents

N,O-Hemiacetal 1	Ph ₃ P=CHR ³	Time (h)	Product	Yield (%) ^a
1a	R ³ =COOMe	6	2a	52
1b		9	2b	53
1c		2	2c	73
1d		2	2d	76
1e		6	2e	76
1c	R^3 =COOMe	3	3c	68
	$R^3 = COPh$	1 ^b	4c	66
1e	$R^3 = COMe$	2	3e	74
	$R^3 = COPh$	2	4e	87
	$R^3=CN$	8	5e	63

A mixture of bicyclo[4.1.0]heptan-7-one N,O-hemiacetal 1 (5 mmol), Wittig reagent (6 mmol), and benzoic acid (0.02 g) in toluene (25 mL) was heated under reflux conditions for 2–9 h.

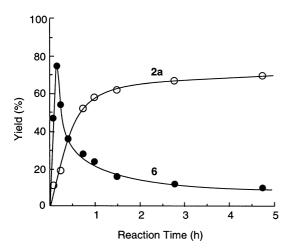


Figure 1. Time-course of the reaction of bicyclo[4.1.0]heptan-7-one *N,O*-hemiacetal **1a** (5 mmol) with Wittig reagent (R³=COOMe, 6 mmol) in the presence of benzoic acid (0.02 g) in refluxing toluene.

reported that the C=C double bond of methylenecyclopropane derivatives carrying an electron-withdrawing group is reactive as a Michael acceptor. ¹⁵ Actually, a reaction of the alkylidene derivative **6**, independently prepared from hemiacetal **7** and Wittig reagent (R³=COOMe), with dimethyl amine readily afforded **2a** (Scheme 3).

The stereochemical configuration of the products was established by comparison of their spectral data with those of authentic samples of 7-endo- and 7-exo-morpholinobicyclo[4.1.0]heptane derivatives which were prepared independently. The endo-amino isomer 9 was obtained by the reaction of bicyclo[4.1.0]heptan-7-one N,O-acetal 8 with trimethylsiloxypropene in the presence of BF₃ etherate, using our previously reported method¹¹ (Scheme 4). On the other hand, the exo-amino isomer 3e was prepared by an addition of morpholine to ring-fused alkylidenecyclopropane 6 (R³=COMe), which was prepared by Wittig reaction of 2-oxopropylidenetriphenylphosphorane with 7-methoxybicyclo[4.1.0]heptan-7-ol^{13a,b} (Scheme 4). The spectral data of product 3e obtained in the present reaction were identical with those of the authentic exo-amino isomer **3e**, and the ¹H NMR spectra of methylene protons of the morpholino ring displayed an AA'XX' pattern, supporting the *exo*-morpholino configuration. ^{16,17}

Proposed reaction pathways of the present reaction are shown in Scheme 5. The formation of ring-fused alkylidenecyclopropanes 6 proceeds probably via iminium cation 10, which is formed from bicyclo[4.1.0]heptan-7-one N,Ohemiacetals 1 under mild acidic conditions. Iminium cation 10 is sufficiently stable in situ to be trapped by various nucleophiles. 6,10 The reaction of 10 with a Wittig reagent gives 1:1 adducts 11, and then an elimination of an iminium salt of triphenylphosphine 12 gives 6. The iminium salt of triphenylphosphine 12 is converted into triphenylphosphine oxide 13 and amine 14 by reaction with water, which is produced initially during the process of formation of iminium cation 10. The liberated amine 14 immediately attacks the β -carbon of α,β -unsaturated derivative **6** from a sterically less-hindered exo-position 10 to give 2-5 predominantly, because the C=C double bond of methylenecyclopropane derivatives 6 carrying an electronwithdrawing group is reactive as a Michael acceptor.¹⁵

^a Isolated yields based on bicyclo[4.1.0]heptan-7-one N,O-hemiacetal 1 employed.

b Refluxed in benzene for 1 h.

Scheme 3.

OSiMe₃
OMe
$$CH_2 = C$$

$$Me$$

$$BF_3 \cdot Et_2O \text{ in } CH_2CI_2$$

$$-78^{\circ}C$$

$$9$$

Scheme 4.

as a solvent. Chemical shifts are given in ppm downfield (δ) from TMS as an internal standard. IR spectra were obtained from neat films unless otherwise noted. MS spectra were obtained at an ionization potential of 70 eV.

3.2. General procedure for preparation of 7-exo-amino-7-endo-substituted bicyclo[4.1.0]heptanes (2-5)

A mixture of bicyclo[4.1.0]heptan-7-one *N,O*-hemiacetal **1** (5 mmol), triphenylphosphorane (6 mmol), and a catalytic amount of benzoic acid (0.02 g) in 25 mL of toluene was stirred under reflux conditions for 2–9 h. The mixture was evaporated in vacuo, and the residue was extracted with pentane (20 mL×5). The combined pentane extracts were concentrated and column-chromatographed on alumina (ether–hexane).

Scheme 5.

3. Experimental

3.1. General

Wittig reagents were prepared as described in the literature. ^{18–20} Bicyclo[4.1.0]heptan-7-one *N,O*-hemiacetals **1a**–**e** were prepared using our previously reported method. ^{11,12} Unless otherwise noted, ¹H and ¹³C NMR spectra were recorded at 270 and 67.5 MHz, respectively, using CDCl₃

3.2.1. Methyl 2-(7-exo-dimethylaminobicyclo[4.1.0]hept7-yl)acetate (2a). IR ν 1740 cm $^{-1}$; 1 H NMR δ 0.95–1.02 (m, 2H), 1.02–1.35 (m, 4H), 1.45–1.58 (m, 2H), 1.84–2.00 (m, 2H), 2.28 (s, 6H), 2.48 (s, 2H), 3.68 (s, 3H); 13 C NMR δ 19.30 (CH $_{2}$), 21.55 (CH $_{2}$), 22.57 (CH), 27.80 (CH $_{2}$), 40.38 (CH $_{3}$), 48.05 (C), 51.48 (OCH $_{3}$), 173.84 (CO); MS m/z (relative intensity) 211 (M $^{+}$, 48), 196 (82), 156 (95), 138 (100), 114 (88). Anal. Calcd for C $_{12}$ H $_{21}$ NO $_{2}$: C, 68.21; H, 10.02; N, 6.63. Found: C, 68.19; H, 10.12; N, 6.49.

- **3.2.2.** Methyl 2-(7-exo-diethylaminobicyclo[4.1.0]hept-7-yl)acetate (2b). IR ν 1740 cm⁻¹; ¹H NMR δ 0.99 (t, J=7.3 Hz, 6H), 0.95–1.05 (m, 2H), 1.05–1.36 (m, 4H), 1.46–1.60 (m, 2H), 1.81–1.99 (m, 2H), 2.46 (s, 2H), 2.59 (q, J=7.3 Hz, 4H), 3.66 (s, 3H); ¹³C NMR δ 15.15 (CH₃), 18.98 (CH₂), 21.66 (CH₂), 21.93 (CH), 30.01 (CH₂), 45.95 (CH₂), 47.03 (C), 51.36 (OCH₃), 174.09 (CO); MS m/z (relative intensity) 239 (M⁺, 45), 224 (48), 210 (100), 180 (55). Anal. Calcd for C₁₄H₂₅NO₂: C, 70.25; H, 10.53; N, 5.85. Found: C, 70.48; H, 10.68; N, 5.65.
- **3.2.3. Methyl 2-(7-exo-pyrrolidinobicyclo[4.1.0]hept-7-yl)acetate (2c).** Bp 108°C (13 mmHg); IR ν 1740 cm⁻¹; H NMR (90 MHz) δ 0.91–1.10 (m, 2H), 1.10–1.46 (m, 4H), 1.46–1.76 (m, 6H), 1.76–2.18 (m, 2H), 2.50 (s, 2H), 2.51–2.71 (m, 4H), 3.66 (s, 3H); ¹³C NMR (22.4 MHz) δ 19.46 (CH₂), 20.73 (CH), 21.87 (CH₂), 23.78 (CH₂), 29.65 (CH₂), 43.21 (C), 47.48 (CH₂), 51.35 (OCH₃), 173.80 (CO); MS m/z (relative intensity) 237 (M⁺, 43), 208 (26), 182 (54), 178 (100), 164 (52). Anal. Calcd for C₁₄H₂₃NO₂: C, 70.85; H, 9.77; N, 5.90. Found: C, 70.80; H, 9.77; N, 5.86.
- **3.2.4.** Methyl **2-**(7-*exo*-piperidinobicyclo[**4.1.0**]hept-7yl)acetate (**2d**). IR ν 1740 cm⁻¹; 1 H NMR δ 0.91–1.01 (m, 2H), 1.01–1.57 (m, 12H), 1.81–1.99 (m, 2H), 2.47 (s, 2H), 2.45–2.55 (m, 4H), 3.67 (s, 3H); 13 C NMR δ 19.37 (CH₂), 21.64 (CH₂), 22.18 (CH), 24.67 (CH₂), 26.78 (CH₂), 28.72 (CH₂), 48.36 (C), 49.76 (CH₂), 51.38 (OCH₃), 173.94 (CO); MS m/z (relative intensity) 251 (M⁺, 25), 194 (39), 192 (100), 178 (34). Anal. Calcd for C₁₅H₂₅NO₂: C, 71.67; H, 10.02; N, 5.57. Found: C, 71.61; H, 10.05; N, 5.80.
- **3.2.5. Methyl 2-(7-exo-morpholinobicyclo[4.1.0]hept-7-yl)acetate (2e).** Mp 41–43°C (pentane, -30°C); IR (KBr) ν 1744 cm⁻¹; ¹H NMR δ 0.95–1.05 (m, 2H), 1.05–1.34 (m, 4H), 1.41–1.55 (m, 2H), 1.82–1.99 (m, 2H), 2.51 (s, 2H), 2.58 (t, J=4.6 Hz, 4H), 3.55 (J=4.6 Hz, 4H), 3.69 (s, 3H); ¹³C NMR δ 19.32 (CH₂), 21.58 (CH₂), 21.66 (CH), 28.81 (CH₂), 47.80 (C), 49.08 (CH₂), 51.57 (OCH₃), 67.76 (CH₂), 173.53 (CO); MS m/z (relative intensity) 253 (M⁺, 54), 194 (100), 180 (73). Anal. Calcd for C₁₄H₂₃NO₃: C, 66.37; H, 9.15; N, 5.53. Found: C, 66.21; H, 8.97; N, 5.43.
- **3.2.6.** 1-(7-exo-Pyrrolidinobicyclo[4.1.0]hept-7-yl)-2-propanone (3c). IR ν 1715 cm⁻¹; ¹H NMR (90 MHz) δ 0.85–1.38 (m, 6H), 1.38–1.69 (m, 6H), 1.69–2.15 (m, 2H), 2.21 (s, 3H), 2.38–2.65 (m, 4H), 2.57 (s, 2H); ¹³C NMR (22.4 MHz) δ 17.92 (CH), 19.42 (CH₂), 21.75 (CH₂), 23.41 (CH₂), 29.56 (CH₃), 40.92 (CH₂), 43.78 (C), 47.77 (CH₂), 208.37 (CO); MS m/z (relative intensity) 221 (M⁺, 8), 178 (100). Anal. Calcd for C₁₄H₂₃NO: C, 75.97; H, 10.47; N, 6.33. Found: C, 75.76; H, 10.54; N, 6.22.
- **3.2.7. 1-**(7-*exo*-Morpholinobicyclo[**4.1.0**]hept-7-yl)-2-propanone (**3e**). Mp 59–61°C (hexane, -30°C); IR (KBr) ν 1713 cm⁻¹; ¹H NMR δ 0.97–1.15 (m, 4H), 1.15–1.45 (m, 4H), 1.80–2.00 (m, 2H), 2.23 (s, 3H), 2.48 (t, J=4.6 Hz, 4H), 2.61 (s, 2H), 3.55 (t, J=4.6 Hz, 4H); ¹³C NMR δ 19.54 (CH₂), 20.38 (CH), 21.64 (CH₂), 30.42 (CH₃), 38.13 (CH₂), 47.51 (C), 49.44 (CH₂), 67.58 (CH₂), 207.49 (CO); MS m/z (relative intensity) 237 (M⁺, 6), 194 (100). Anal. Calcd for C₁₄H₂₃NO₂: C, 70.85; H, 9.77; N, 5.90. Found: C, 71.00; H, 9.65; N, 5.88.

- **3.2.8. 2-(7-***exo-***Pyrrolidinobicyclo**[**4.1.0**]**hept-7-yl)acetophenone** (**4c**). IR ν 1693 cm⁻¹; 1 H NMR δ 0.95–1.40 (m, 6H), 1.40–1.69 (m, 6H), 1.85–2.01 (m, 2H), 2.44–2.60 (m, 4H), 3.17 (s, 2H), 7.39–7.59 (m, 3H), 7.95–8.05 (m, 2H); 13 C NMR δ 19.75 (CH₂), 20.36 (CH), 21.71 (CH₂), 23.13 (CH₂), 32.28 (CH₂), 42.64 (C), 48.09 (CH₂), 127.71 (CH), 128.34 (CH), 132.45 (CH), 137.72 (C), 198.81 (CO); MS m/z (relative intensity) 212 (M⁺–pyrrolidine, 100), 183 (80). Anal. Calcd for C₁₉H₂₅NO: C, 80.52; H, 8.89; N, 4.94. Found: C, 80.43; H, 9.10; N, 4.83.
- **3.2.9. 2-(7-***exo***-Morpholinobicyclo[4.1.0]hept-7-yl)acetophenone** (**4e**). Mp 78–79°C (hexane); IR (KBr) ν 1690 cm⁻¹; ¹H NMR δ 0.95–1.34 (m, 6H), 1.41–1.58 (m, 2H), 1.85–2.19 (m, 2H), 2.54 (t, J=4.6 Hz, 4H), 3.18 (s, 2H), 3.52 (t, J=4.6 Hz, 4H), 7.41–7.61 (m, 3H), 7.95–8.05 (m, 2H); ¹³C NMR δ 19.88 (CH₂), 21.37 (CH), 21.73 (CH₂), 31.95 (CH₂), 47.71 (C), 49.85 (CH₂), 67.64 (CH₂), 127.85 (CH), 128.57 (CH), 132.79 (CH), 137.61 (C), 198.71 (CO); MS m/z (relative intensity) 299 (M⁺, 7), 282 (17), 194 (100). Anal. Calcd for C₁₉H₂₅NO₂: C, 76.22; H, 8.42; N, 4.68. Found: C, 76.18; H, 8.46; N, 4.64.
- **3.2.10. 2-(7-exo-Morpholinobicyclo[4.1.0]hept-7-yl)-acetonitrile (5e).** Mp 91–92°C (hexane); IR (KBr) ν 2245 cm⁻¹; ¹H NMR δ 1.01–1.40 (m, 6H), 1.40–1.60 (m, 2H), 1.85–2.05 (m, 2H), 2.48 (s, 2H), 2.71 (t, J=4.6 Hz, 4H), 3.60 (t, J=4.6 Hz, 4H); ¹³C NMR δ 11.39 (CH₂), 18.51 (CH₂), 21.37 (CH₂), 21.65 (CH), 46.88 (C), 48.95 (CH₂), 67.38 (CH₂), 119.64 (CN); MS m/z (relative intensity) 220 (M⁺, 43), 180 (47), 165 (49), 140 (100). Anal. Calcd for C₁₃H₂₀N₂O: C, 70.87; H, 9.15; N, 12.72. Found: C, 71.11; H, 8.90; N, 12.44.

3.3. Preparation of 7-endo-amino-7-exo-substituted bicyclo[4.1.0]heptane 9

To a magnetically stirred solution of bicyclo[4.1.0]heptan-7-one N,O-acetal **8** (2.11 g, 10 mmol) and 2-trimetylsiloxy-propene (1.95 g, 15 mmol) in CH_2Cl_2 (30 mL), $BF_3 \cdot Et_2O$ (1.42 g, 10 mmol) was added via syringe over a period of 10 min at $-78^{\circ}C$. The reaction mixture was stirred for 3 h at this temperature, allowed to warm to $0^{\circ}C$ over a period of 2 h, and quenched with saturated NaCl aqueous solution. The organic layer was separated, and the aqueous layer was extracted with CH_2Cl_2 (10 mL×2). The combined organic layers were dried over Na_2SO_4 , and evaporated. The residue was column-chromatographed on silica gel (ether–hexane, 2:1) to give **9** as an oil (1.32 g, 56% yield), which was solidified on standing. The analytical sample was obtained by recrystallization from hexane at $-30^{\circ}C$.

3.3.1. 1-(7-endo-Morpholinobicyclo[4.1.0]hept-7-yl)-2-propanone (9). Mp 60–61°C (hexane, -30°C); IR (KBr) ν 1707 cm⁻¹; ¹H NMR δ 0.82–0.95 (m, 2H), 1.12–1.31 (m, 2H), 1.40–1.65 (m, 4H), 1.65–1.90 (m, 2H), 2.18 (s, 3H), 2.49 (s, 2H), 2.51–2.70 (m, 4H), 3.48–3.68 (m, 2H), 3.68–3.85 (m, 2H); ¹³C NMR δ 19.25 (CH₂), 21.19 (CH), 22.01 (CH₂), 30.89 (CH₃), 44.94 (C), 45.95 (CH₂), 50.30 (CH₂), 67.39 (CH₂), 208.64 (CO); MS m/z (relative intensity) 237 (M⁺, 19), 220 (31), 195 (53), 194 (100). Anal. Calcd for

C₁₄H₂₃NO₂: C, 70.85; H, 9.77; N, 5.90. Found: C, 70.93; H, 9.80; N, 5.87.

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References

- 1. Salaun, J. Chem. Rev. 1983, 83, 619-632.
- Wasserman, H. H.; Dion, R. P. Tetrahedron Lett. 1982, 23, 1413–1416.
- 3. van Tilborg, W. J. M.; Steinberg, H.; de Boer, Th. J. *Recl. Trav. Chim.-Pays Bas* **1974**, *93*, 290–293.
- Wasserman, H. H.; Dion, R. P. Tetrahedron Lett. 1982, 23, 785–788.
- van Tilborg, W. J. M.; Dooyewaard, G.; Steinberg, H.; de Boer, Th. J. *Tetrahedron Lett.* 1972, 1677–1680.
- Wasserman, H. H.; Baird, M. S. Tetrahedron Lett. 1971, 3721–3724.
- 7. de Boer, Th. J. Angew. Chem. 1972, 84, 303-304.

- 8. Wasserman, H. H. Angew. Chem. 1972, 84, 312-313.
- 9. Vilsmaier, E.; Klein, C. M. Synthesis 1981, 206-207.
- Vilsmaier, E.; Stamm, T.; Michels, G. Synthesis 1988, 858– 862
- 11. Chiba, T.; Saitoh, I.; Okimoto, M. Synthesis **1999**, 1022–1026
- Chiba, T.; Saitoh, I.; Okimoto, M.; Tanase, T.; Yano, S. J. Org. Chem. 1999, 64, 2516–2519.
- Benzoic acid is known to catalyze the reaction of cylopropanone hemiacetals with Wittig reagents; (a) Osborne, N. F. *J. Chem. Soc.*, *Perkin Trans. 1* 1982, 1435–1439.
 (b) Spitzner, D.; Swoboda, H. *Tetrahedron Lett.* 1986, 27, 1281–1284. (c) Kortmann, I.; Westermann, B. *Synthesis* 1995, 931–933.
- Chowdhury, M. A.; Senboku, H.; Tokuda, M.; Masuda, Y.;
 Chiba, T. *Tetrahedron Lett.* **2001**, *42*, 7075–7078.
- 15. Liese, T.; Teichmann, S.; de Meijere, A. Synthesis 1988, 25-
- 16. Vilsmaier, E.; Tröger, W. Angew. Chem. 1979, 91, 860–861.
- 17. Vilsmaier, E.; Tröger, W.; Haag, G. *Chem. Ber.* **1981**, *114*, 67–79.
- Isler, O.; Gutmann, H.; Montavon, M.; Rüegg, R.; Ryser, G.;
 Zeller, P. Helv. Chim. Acta 1957, 40, 1242–1249.
- 19. Ramirez, F.; Dershowitz, S. J. Org. Chem. 1957, 22, 41-45.
- 20. Trippett, S.; Walker, D. M. J. Chem. Soc. 1961, 1266-1272.